## METHODOLOGY





Simultaneous untargeted and targeted metabolomics profiling of underivatized primary metabolites in sulfur-deficient barley by ultra-high performance liquid chromatography-quadrupole/time-of-flight mass spectrometry

Hikmat Ghosson<sup>1,2</sup>, Adrián Schwarzenberg<sup>1\*</sup>, Frank Jamois<sup>1</sup> and Jean-Claude Yvin<sup>1</sup>

## Abstract

**Background:** Metabolomics based on mass spectrometry analysis are increasingly applied in diverse scientific domains, notably agronomy and plant biology, in order to understand plants' behaviors under different stress conditions. In fact, these stress conditions are able to disrupt many biosynthetic pathways that include mainly primary metabolites. Profiling and quantifying primary metabolites remain a challenging task because they are poorly retained in reverse phase columns, due to their high polarity and acid–base properties. The aim of this work is to develop a simultaneous untargeted/targeted profiling of amino acids, organic acids, sulfur metabolites, and other several metabolites. This method will be applied on sulfur depleted barley, in order to study this type of stress, which is difficult to detect at early stage. Also, this method aims to explore the impact of this stress on barley's metabolome.

**Results:** Ultra-high performance liquid chromatography–high resolution mass spectrometry-based method was successfully applied to real samples allowing to discriminate, detect, and quantify primary metabolites in short-runs without any additional sampling step such as derivatization or ion pairing. The retention of polar metabolites was successfully achieved using modified C18 columns with high reproducibility (relative standard deviation below 10%). The quantification method showed a high sensitivity and robustness. Furthermore, high resolution mass spectrometry detection provided reliable quantification based on exact mass, eliminating potential interferences, and allowing the simultaneous untargeted metabolomics analysis. The untargeted data analysis was conducted using Progenesis QI software, performing alignment, peak picking, normalization and multivariate analysis. The simultaneous analysis provided cumulative information allowing to discriminate between two plant batches. Thus, discriminant biomarkers were identified and validated. Simultaneously, quantification confirmed coherently the relative abundance of these biomarkers.

**Conclusions:** A fast and innovated simultaneous untargeted/targeted method has successfully been developed and applied to sulfur deficiency on barley. This work opens interesting perspectives in both fundamental and applied research. Biomarker discovery give precious indication to understand plant behavior during a nutritional deficiency.

\*Correspondence: adrian.schwarzenberg@roullier.com

<sup>1</sup> Centre Mondial de l'Innovation Roullier (CMI), 18 Avenue Franklin

Roosevelt, 35400 Saint-Malo, France

Full list of author information is available at the end of the article



© The Author(s) 2018. This article is distributed under the terms of the Creative Commons Attribution 4.0 International License (http://creativecommons.org/licenses/by/4.0/), which permits unrestricted use, distribution, and reproduction in any medium, provided you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The Creative Commons Public Domain Dedication waiver (http://creativecommons.org/ publicdomain/zero/1.0/) applies to the data made available in this article, unless otherwise stated.

Thus, direct or indirect measurement of these compounds allows a real time fertilization management and encounter the challenges of sustainable agriculture.

Keywords: Plant metabolomics, Primary metabolites, Organic acids, Amino acids, LC–HRMS, Sulfur deficiency

## Background

Metabolomics based on gas chromatography–mass spectrometry (GC–MS) and/or liquid chromatography–mass spectrometry (LC–MS) are widely applied for exhaustive and specified studies in many different scientific fields [1]. Both untargeted and targeted strategies are being developed, notably in agronomy and plant biology [2].

Many different metabolomics studies in plant biology are emerging for various aims (e.g. discovering new biocontrol agents, phytomedicine, etc.). Otherwise, in order to understand plant physiological reactions and behaviors under different biotic and/or abiotic stress (e.g. drought stress, nutrient deficiencies, bio-stimulant applications, microorganism effects, associated crop), high-throughput methods need to be developed. This is to identify and/or quantify involved biomarkers in a complex matrix. The different types of stress or treatments are able to modify and disrupt biosynthesis pathways [3-8]. Furthermore, biomolecules implied in these pathways, mainly primary and polar metabolites, are present in low concentrations, leading to many difficulties during the extraction and chromatographic separation. The high polarity and acid-base properties implies time consuming sample preparation (e.g. derivatization). Moreover, delicate chromatographic optimization should be achieved, in order to assure reliable and robust separation and detection with GC-MS and/or LC-MS systems.

Several targeted methods based on LC–MS have been reported in the literature [4, 9–22], mainly for primary metabolites analysis [9–22]. However, most of LC–MS methods needed additional sampling steps [15, 18–20], or were performed using hydrophilic interaction liquid chromatography (HILIC) [4, 12, 21, 22]. HILIC methods required a delicate, careful optimization, and relatively longtime column conditioning. On the other hand, ion pairing technique was also applied [9, 16, 17, 22], however it risks the residual system contamination. Liu et al. [13] reported a targeted high performance liquid chromatography–mass spectrometry (HPLC–MS) method that allowed to analyze 28 polar metabolites in 25 min without additional sampling steps.

Alternatively, several untargeted GC–MS based methods applied in plants have been reported [2, 3, 6, 23–26]. GC–MS based techniques are mainly utilized for primary metabolites profiling [23, 27], providing high chromatographic resolution and high sensitivity.

Moreover, the high reproducibility of electronic impact (EI) fragmentation provides reliable metabolites identification [27]. Nevertheless, samples required derivatization to analyze these non-volatile and polar metabolites. Furthermore, thermolabile metabolites analysis are difficult due to thermo-degradation [27].

Hence, untargeted LC–MS based methods are rapidly developing lately [28]. Due to their performance to analyze wide variety of metabolites [27], many LC–MS based methods were recently developed and increasingly applied for metabolomics profiling in plants [23, 29–33] and phytomedicine [34, 35].

It is worth to mention that simultaneous targeted and untargeted metabolomics was applied in a phytomedicine study [36]. However, none of the reported studies showed this approach on the study of primary metabolites in plant metabolomics. Our objective was to develop, validate and apply an ultra-high performance liquid chromatography-high resolution mass spectrometry (UPLC-HRMS) based metabolomics method for analyzing underivatized primary metabolites in less than 10 min.

As primary metabolites play an essential role in plant growth, development, and reproduction, and secondary metabolites as flavonoids and polyphenols are involved in plant defense [37, 38], their abundances explain an important part of plant behaviors under different biotic or abiotic stress [3–8, 27]. Consequently, 34 major metabolites were selected from different biochemical pathways in order to explain the physiological behavior under nutrient deprival; sulfur-deficiency in our case.

In fact, sulfur deficiency can lead to yield losses due to its non-visual symptoms, it is not easily identifiable because of the confusion between sulfur deficiency and nitrogen deficiency [39]. Moreover, early stage sulfur deficiency is also difficult to be detected, due to the usual less accurate prediction when the sampling is on the early growth stage. However, analyzing biochemical indicators as glutathione can lead to more reliable diagnosis [40].

The workflow represented in Fig. 1 consists of extracting metabolites from plant material, realizing LC–HRMS analysis, and processing same data with two different approach: untargeted profiling for batch discrimination and biomarkers determination, and targeted quantification and biomarkers identification.



## Methods

#### Chemicals and reagents

Amino acids, and organic acids HPLC quality standards proline (Pro), isoleucine (Ile), leucine (Leu), asparagine (Asn), aspartic acid (Asp), glutamine (Gln), glutamic acid (Glu), lysine (Lys), methionine (Met), histidine (His), phenylalanine (Phe), arginine (Arg), tyrosine (Tyr), tryptophan (Trp), 5 amino acid derivatives: O-acetyl-serine, thiamine, glutathione reduced (GSH), S-adenosyl-methionine (SAM), glutathione oxidized (GSSG), fumaric acid, succinic acid, malic acid, phospho(enol)pyruvic acid, cis-aconitic acid, shikimic acid, citric acid, isocitric acid, gluconic acid, 2 phosphorylated sugars: D-glucose 6-phosphate, trehalose 6-Phosphate, 4 secondary metabolites: gallic acid, azelaic acid, kaempferol, chlorogenic acid, and two internal standards: Lactitol and Taurine, were purchased from Sigma-Aldrich (Saint Quentin, France).

Ultra-pure water was prepared by a Milli-Q Advantage A10 system (Darmstadt, Germany), Acetonitrile (ACN) and Methanol (MeOH) Optima LC–MS grade were purchased from Fisher (Leicester, UK), Formic Acid (FA) LC–MS grade and perchloric acid (PCA) were purchased from Merck (Darmstadt, Germany).

## **Plant materials**

Method was applied on barley (Hordeum vulgare) plants grown and treated in Centre Mondial de l'Innovation Roullier greenhouse (Saint-Malo, Bretagne, France). Seeds of Hordeum vulgare cv. Irina were germinated on vermiculate for 3 days in the dark and for additional 4 days under light conditions. After 1 week, seedlings were transplanted to a 5.9 L tank in greenhouse that was set to a 14/10 h day/night cycle at a day/night temperature of 28/25 °C with 40-50% relative humidity. Plants were divided into two batches: (1) S-sufficient (0.5 mM) plants were grown with a complete nutritive solution, (2) S-deficient plants were grown in a different solution with a low concentration of S (0 mM):  $Ca(NO_3)^{2-}$  2 mM,  $K_2HPO_4$  1 mM,  $MgCl_2$ 0.5 mM,  $NH_4H_2PO_4$  0.5 mM,  $CaCl_2$  0.5 mM,  $H_3BO_3$ 0.001 mM, MnNO<sub>3</sub> 0.0025 mM, ZnNO<sub>3</sub> 0.0005 mM, CuNO<sub>3</sub> 0.0002 mM, (NH<sub>4</sub>)<sub>6</sub>Mo<sub>7</sub>O<sub>24</sub> 0.00001 mM and EDTA, 2NaFe 0.1 mM. The nutrient solution was buffered to pH 5.9 and renewed every 2 days and continuously aerated. After 2 weeks of stress, leaves and roots of two batches were harvested and immediately frozen in liquid nitrogen and then stored at -80 °C until analysis. Before extraction, materials were grinded in CryoMill (5 µm, Retsch, Haan, Germany).

#### **UPLC conditions**

Ultra-high performance liquid chromatography analysis was performed using a Waters Acquity H-Class UPLC system (Waters Corp, Milford, USA). Amino acids and sulfur contain metabolites separation was performed using a Waters UPLC HSS T3 column ( $2.1 \times 100$  mm,  $1.8 \mu$ m). The mobile phase consisting of water containing 0.1% formic acid (A) and acetonitrile/methanol 50:50 v/v containing 0.1% formic acid (B) was applied with the optimized gradient elution as follows: 100% A at 0–1.5 min, 100–80% A at 1.5–2 min, 80–20% A at 2–2.5 min, 20% A at 2.5–4.5 min, 20–100% A at 4.5–5 min, 100% A at 5–7 min. The flow rate was kept at 0.4 mL/min, and column temperature was maintained at 25 °C.

The separation of organic acids, phosphorylated sugars, secondary metabolites and two amino acids (aspartic acid and glutamic acid) was achieved using a Phenomenex Luna<sup>®</sup> Omega PS C18 ( $100 \times 2.1 \mu$ m,  $1.6 \mu$ m) column (Torrance, USA). The mobile phase consisting of water containing 0.5% formic acid (A) and methanol/water 70:30 v/v containing 0.5% formic acid (B) was applied with the optimized gradient elution as follows: 100% A at 0–1 min, 100–20% A at 1–4 min, 20–0% A at 4–6.5 min, 0% A at 6.5–7.5 min, 0–100% A at 7.5–7.9 min, 100% A at 7.9–10 min. The flow rate was kept at 0.3 mL/min, column temperature was maintained at 35 °C. The injection volume for both columns was 10 µL and samples were maintained at 10 °C.

## QToF conditions

High resolution mass spectrometry detection of metabolites was performed by Waters Xevo G2-S quadrupole/ time-of-flight mass spectrometer (QToF MS) (Waters Corp, Milford, USA) equipped with an electrospray ionization (ESI) source. For positive ESI, source voltage was set to 0.5 kV and cone voltage was 15 V, whereas source temperature was maintained at 130 °C with a cone gas flow of 20 L/h. Desolvation temperature was at 500 °C with desolvation gas flow of 800 L/h. For negative ESI, source voltage was set to 2.5 kV and cone voltage was 30 V, whilst source temperature was maintained at 130 °C with a cone gas flow of 20 L/h, desolvation temperature was at 550 °C with desolvation gas flow of 900 L/h. Leucine-Enkephalin (Waters, Manchester, UK) was used as lockmass reference, (ion at m/z 556.2771 in positive mode and m/z 554.2615 in negative mode), which was introduced by a Lockspray at 10 µL/min for real-time data calibration. The MS<sup>E</sup> data were acquired in centroid mode using a scan range 50-800 Da, scan time 0.1 s, resolution was set at 20000 full width half maximum (FWHM), and a collision energy ramp 40–80 V.

Molecular ions  $[M+H]^+$  and  $[M-H]^-$  were detected in positive and negative ionization, respectively. Chromatographic peaks were extracted from the full scan chromatograms using MassLynx V4.1 software (Waters Inc., USA), based on  $[M+H]^+$  and  $[M-H]^-$  ions. Peak areas were integrated using TargetLynx software (Waters Inc., USA), and data treatment for untargeted analysis were performed by Progenesis QI software (Nonlinear Dynamics, Newcastle, UK) and EZinfo 3.0 software (Umetrics AB, Umeå, Sweden).

## **Method validations**

Determination of the limit of detection (LOD), limit of quantitation (LOQ), and linearity were carried out using a series of diluted mixed standards of metabolites. The concentrations were chosen through preliminary tests to establish the linear range and enable quantification in the plant material of interest.

To determine the method precision, three concentration levels (one close to LOQ, one intermediate and one close to the upper limit of linear range) of mixed standards were injected ten times. Otherwise, different plant samples were injected ten times for intra-sample validation, also 4 biological replicate samples were analyzed for intra-day, and inter-day validation within 6 months. The repeatability and reproducibility intra-sample, intraday and inter-day for each compound were estimated by calculation of the respective relative standard deviation (RSD) values (Additional file 1: Tables S1–S6).

#### **Extraction method**

20 mg of frozen grinded fresh leaves and roots were weighted in a 2 mL Eppendorf tubes, then 500 µL of cold water/methanol 70:30 v/v (-20 °C) containing 0.4% of perchloric acid (v/v) solvent were added. Samples were shaken with vortex for 10 min. Then, they were centrifuged using an Eppendorf Centrifuge 5427 R (Hamburg, Germany) for 15 min 12,700 RPM at 4 °C. Supernatants were collected and introduced in a new 2 mL Eppendorf tubes. A second extraction was performed adding 500 µL of water +0.1% perchloric acid (v/v) to leaves and roots, shaken for 5 min with vortex, and centrifuged for 15 min with 12,700 rotation per minute (RPM) on 4 °C. Then supernatants were recuperated in same tubes of the first extraction. Supernatants were mixed and centrifuged for 10 min in order to eliminate suspended particles without introducing contaminants issued from filters. Finally, supernatants were diluted 2 times with water +0.1% Formic acid (v/v) and introduced in 2 mL LC–MS vials.

Perchloric acid was used to protect metabolites from enzymatic degradation [41], and to avoid sulfur metabolites oxidation and degradation [14, 42, 43] under basic and neutral condition. Formic acid was used for enhancing electrospray ionization.

## Data treatment

LC–HRMS<sup>E</sup> acquired data were treated in two different paths: untargeted analysis and targeted analysis.

Untargeted data analysis was performed using Progenesis QI software. Data were processed in successive treatment steps as peak alignment, peak picking and normalization to obtain data matrix. This matrix was used to perform multivariate analysis.

Targeted analysis was performed using TargetLynx software. Targeted metabolites' m/z ratios were extracted from chromatogram, and chromatographic peaks were integrated to quantify metabolites using calibration curves with internal standards correction. Chosen internal standards were Taurine for positive ionization mode and Lactitol for negative ionization mode.  $[M+H]^+$  and  $[M-H]^-$  ions were used for quantification.

#### Method application

The method was evaluated on barley under sulfur controlled and deficient conditions. Leaves and roots were analyzed separately for untargeted/targeted metabolite profiling. The data issued from the LC–HRMS analysis were used to perform the untargeted data analysis and targeted quantification (Fig. 1). Quality control solution (QC) was prepared by mixing similar volume aliquots from all samples. QC solution was prepared in order to obtain the variability of all samples.

Analytical sequence consisted in a calibration curve followed by 10 consecutive injections of QC to stabilize the LC system. Then, all samples were injected randomly to minimize the effect of instrumental drift. A QC was injected every 5 samples as well as a standard QC in order to control carry over, stability and robustness.

### **Results and discussion**

## Optimization and validation of targeted profiling method Chromatographic separation

Amino acids and sulfur containing metabolites separation was performed using an HSS T3 column. The majority of compounds (13 out of the 17 compounds) were chromatographically separated (Fig. 2). Structural isomers Isoleucine and Leucine are chromatographically resolved as shown in Fig. 4. Co-eluting compounds as Histidine and Arginine can be differentiated by their mass difference. The HSS T3 column contains a modified C18 stationary phase with 100% silica base, which provides a hydrophilic interaction with polar metabolites, allowing to enhance their retention. Furthermore, this 100% silica base allows to use 100% aqueous eluting phase, so polar metabolites are weakly eluting.





**Fig. 3** Extracted ion chromatogram of organic acids—Phenomenex Luna<sup>®</sup> Omega PS C18 column—ESI: 1, Fumaric acid; 2, Succinic acid; 3, Aspartic acid; 4, Malic acid; 5, Glutamic acid; 6, Phospho(enol)pyruvic acid; 7, Gallic acid; 8, Cis-Aconitic acid; 9, Shikimic acid; 10, Azelaic acid; 11, Citric acid; 12, Isocitric acid; 13, Gluconic acid; 14, D-Glucose 6-Phosphate/Fructose-6-Phosphate; 15, Kaempferol; 16, Chlorogenic acid; 17, Trehalose 6-Phosphate



Organic acids and other metabolites including phosphorylated sugars, secondary metabolites and two amino acids aspartate and glutamate were separated using a Luna<sup>®</sup> Omega PS C18 column that allowed to chromatographically resolve 13 of 17 compounds as shown in Fig. 3. The structural isomers isocitrate and citrate were completely separated as represented in Fig. 4, whereas co-eluting analytes as fumarate and malate can be differentiated by their mass difference. However, the phosphorylated sugar D-glucose 6-phosphate was detected but it could not be separated from its isomer D-fructose 6-phosphate. The Luna® Omega PS C18 column is also a modified C18 stationary phase, including positive charge implanting. This positive charge allows strong retention of organic acids, due to the charge interaction with carboxyl function. Additionally, the 100% aqueous eluting phase is applicable with this column.

Detection and quantification were bolstered using high resolution QToF mass spectrometer, which is able to discriminate between metabolites' ions with the same nominal mass (e.g. cis-aconitate and shikimate). High resolution detection also allowed the elimination of potential interferences as isotope contributions. In fact, reliability of quantification by high resolution mass spectrometry is assured by exact mass, due to the elimination of potential errors issued from interferences [44]. Furthermore, HRMS acquisition is necessary to apply untargeted analysis of acquired data. For cis-aconitate, two chromatographic peaks corresponding to its accurate mass (m/z 173.0092) were reported. One of the two peaks corresponds to the same retention time of that of Isocitrate. This peak represents a fragment issued from isocitrate giving an ion with the same elemental composition that the cis-aconitate (water loss) (Additional file 1: Figures S1 and S2).

We have found that amino acids can be detected in both positive mode (as protonated ions) and negative mode (as deprotonated ions). Most of amino acids have shown a better response in positive mode. However, aspartic acid and glutamic acid showed a better signal in negative mode. On the other hand, O-acetyl-serine, methionine, tryptophan, glutathione reduced and glutathione oxidized were more sensitive in positive mode. The introduction of primary metabolites with an m/zbelow 100 (Pyruvate as an example) was difficult due to instrumental limits.

## Limit of detection, limit of quantification and linearity

The LOQ was determined as the smallest amount of a compound reliably quantified showing a signal to noise (S/N) value above 10, and the LOD value is the smallest amount of a compound that can be reliably distinguished, usually showing an S/N value above 3. The linear range was determined using 4 replicates of successively diluted mix of standards.

Positive ionization mode showed a very sensitive detection for amino acid and sulfur metabolites.

Compound	Elemental composition	lons detected	Measured <i>m/z</i>	RT (min)	Linearity range (ng mL <sup>-1</sup> )	R <sup>2</sup>	LOD (ng mL <sup>-1</sup> )	LOQ (ng mL <sup>-1</sup> )
ESI—								
Fumaric acid	$C_4H_4O_4$	$[M - H]^{-}$	115.0037	1.10	200-10,000	0.9995	75	200
Succinic acid	$C_4H_6O_4$	$[M - H]^{-}$	117.0193	1.80	500-10,000	0.9996	150	500
Aspartic acid	C <sub>4</sub> H <sub>7</sub> NO <sub>4</sub>	$[M - H]^{-}$	132.0302	0.80	250-5000	0.9966	15	50
Malic acid	C <sub>4</sub> H <sub>6</sub> O <sub>5</sub>	$[M - H]^{-}$	133.0142	1.10	100-5000	0.9975	15	50
Glutamic acid	C <sub>5</sub> H <sub>9</sub> NO <sub>4</sub>	$[M - H]^{-}$	146.0459	0.82	50-2500	0.9993	15	50
Phospho(enol)pyru- vic acid	$C_3H_5O_6P$	$[M - H]^{-}$	166.9751	1.25	1000-10,000	0.9999	200	500
Gallic acid	C <sub>7</sub> H <sub>6</sub> O <sub>5</sub>	$[M - H]^{-}$	169.0142	3.32	100-5000	0.9997	0.75	2.5
Cis-aconitic acid	C <sub>6</sub> H <sub>6</sub> O <sub>6</sub>	$[M - H]^{-}$	173.0092	3.06	250-5000	0.9969	75	250
Shikimic acid	C <sub>7</sub> H <sub>10</sub> O <sub>5</sub>	$[M - H]^{-}$	173.0455	1.18	100-2500	0.9998	4.5	15
Azelaic acid	C <sub>9</sub> H <sub>16</sub> O <sub>4</sub>	$[M - H]^{-}$	187.0976	6.09	10-1000	0.9998	0.5	1.5
Citric acid	C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>	$[M - H]^{-}$	191.0197	1.63	50-1000	0.9969	5	20
Isocitric acid	C <sub>6</sub> H <sub>8</sub> O <sub>7</sub>	$[M - H]^{-}$	191.0197	1.16	50-1000	0.9983	5	20
Gluconic acid	C <sub>6</sub> H <sub>12</sub> O <sub>7</sub>	$[M - H]^{-}$	195.0510	0.88	50-1000	0.9968	4	15
Kaempferol	C <sub>15</sub> H <sub>10</sub> O <sub>6</sub>	$[M - H]^{-}$	285.0405	7.44	20-1000	0.9951	0.5	2
Chlorogenic acid	C <sub>16</sub> H <sub>18</sub> O <sub>9</sub>	$[M - H]^{-}$	353.0878	4.94	20-1000	0.9978	1.5	5
Trehalose 6-Phos- phate	C <sub>12</sub> H <sub>23</sub> O <sub>14</sub> P	$[M - H]^{-}$	421.0753	1.01	50-1000	0.9955	0.5	1.5
ESI+								
Proline	C <sub>5</sub> H <sub>9</sub> NO <sub>2</sub>	$[M + H]^+$	116.0706	0.77	2.5-125	0.9981	0.75	2.5
Isoleucine	C <sub>6</sub> H <sub>13</sub> NO <sub>2</sub>	$[M + H]^+$	132.1019	2.41	6.5-125	0.9984	2	6.5
Leucine	C <sub>6</sub> H <sub>13</sub> NO <sub>2</sub>	$[M + H]^+$	132.1019	2.58	6.5-125	0.9994	2	6.5
Asparagine	$C_4H_8N_2O_3$	$[M + H]^+$	133.0608	0.85	20-100	0.9972	6	20
Glutamine	$C_5H_{10}N_2O_3$	$[M + H]^+$	147.0764	0.64	10–50	0.9918	0.3	1
Lysine	C <sub>6</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub>	$[M + H]^+$	147.1128	0.54	7.5–150	0.9987	2	7.5
O-Acetyl-serine	C <sub>5</sub> H <sub>9</sub> NO <sub>4</sub>	$[M + H]^+$	148.0604	0.74	10-100	0.9964	5	10
Methionine	C <sub>5</sub> H <sub>11</sub> NO <sub>2</sub> S	$[M + H]^+$	150.0583	1.41	15-150	0.9981	0.5	1.5
Histidine	$C_6H_9N_3O_2$	$[M + H]^+$	156.0768	0.60	8–150	0.9982	0.5	1.5
Phenylalanine	C <sub>9</sub> H <sub>11</sub> NO <sub>2</sub>	$[M + H]^+$	166.0863	3.57	3-200	0.9990	0.15	0.5
Arginine	$C_6H_{14}N_4O_2$	$[M + H]^+$	175.1190	0.60	9–200	0.9968	0.05	0.2
Tyrosine	C <sub>9</sub> H <sub>11</sub> NO <sub>3</sub>	$[M + H]^+$	182.0812	3.20	9–200	0.9986	0.5	2
Tryptophan	C <sub>11</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub>	$[M + H]^+$	205.0972	3.61	2-100	0.9989	0.35	1.2
Thiamine	C <sub>12</sub> H <sub>17</sub> CIN <sub>4</sub> OS	$[M + H]^+$	265.1118	0.83	2-100	0.9999	0.5	2
Glutathione reduced	C <sub>10</sub> H <sub>17</sub> N <sub>3</sub> O <sub>6</sub> S	$[M + H]^+$	308.0911	1.69	5-100	0.9996	0.15	0.5
S-adenosyl-methio- nine	C <sub>15</sub> H <sub>22</sub> N <sub>6</sub> O <sub>5</sub> S	$[M + H]^+$	399.1445	0.74	10-100	0.9974	0.75	2.5
Glutathione oxidized	$C_{20}H_{32}N_6O_{12}S_2$	$[M + H]^+$	613.1592	3.45	10-100	0.9974	0.5	1

## Table 1 Retention times (RT), linearity ranges, LOD and LOQ of polar metabolites tested

Calculated LOQ was 10 ng mL<sup>-1</sup> and lower for all amino acids and sulfur metabolites, as shown in Table 1. Only asparagine showed a LOQ higher than 10 ng mL<sup>-1</sup>.

### Precision

This method showed good reproducibility in retention time and peak area for all standard amino acids and sulfur metabolites detected in positive ionization mode at all injection levels. The overall RSD of ten injections is below 2% for retention time and below 8% for peak area (Additional file 1: Tables S1, S2 and S3).

For ESI–, organic acids, amino acids, phosphorylated sugars and secondary metabolites showed a LOQ between 1.5 and 500 ng mL<sup>-1</sup>. The negative ionization was sensitive enough to detect and quantify all organic acids in plant samples.

The precision in retention time and peak area for standard organic acids, amino acids, phosphorylated

sugars and secondary metabolites detected in negative ionization was better than that of amino acids and sulfur metabolites. The RSD of ten injections is below 0.7% for retention time and below 8% for peak area at lowest injection level, smaller RSD values for peak area are obtained with intermediate and high injection levels (Additional file 1: Tables S4, S5 and S6).

After targeted optimization and validation, the method was applied for simultaneous targeted and untargeted metabolites profiling on real sample.

# Simultaneous untargeted profiling and quantification of discriminant features

The aim of this application was to discriminate between two batches of barley, one batch under controlled sulfur conditions (+S) and a second batch under sulfur deprivation conditions (-S). Studied samples consisted in 8 biological replicates from each batch, leaves and roots were separately analyzed by two columns: the HSS T3 column in positive polarity and Luna<sup>®</sup> Omega PS C18 column in negative polarity.

Principal Components Analysis (PCA) is a descriptive unsupervised discrimination analysis that allows to explain variations between different runs without any a priori knowledge of metabolite profiles. After unsupervised analysis, potential features of two groups were exploited using explicative supervised Orthogonal Projections to Latent Structures Discriminant Analysis (OPLS-DA). Features determination was followed by their identification using injected standard solutions. Also, their relative estimated abundance was compared to targeted quantification when they are included in targeted compounds.

Multivariate analysis was performed using Progenesis QI software. Raw LC-HRMS data generated by the instrument were imported to software without any conversion to perform data analysis. A 2D ion intensity map was generated with the retention time and m/z information as the ordinate and abscissa respectively. Peak alignment was carried out using a QC run as reference. Alignment score values for all runs were higher than 90%. Peak picking threshold of sensitivity was set at 3, and normalization was performed using all compounds. Time limits and adducts used for each analysis are represented in Additional file 1: Tables S7 and S8.

Unsupervised PCA was initially applied based on the ions detected in negative and positive modes and filtered by means of a max fold change  $\geq 2$  and an analysis of variance (ANOVA) p value  $\leq 0.05$  for visualizing the distribution of all the samples. Two-component PCA models accounted of the total variance: 55.14% for leaves and 65.31% for roots in positive ionization, 50.87% for leaves,

Page 9 of 17

and 65.94% for roots in negative ionization as shown in Fig. 5.

PCA demonstrated a difference between the two batches, sulfur deficient samples are regrouped (orange) and represented a notable discrimination to control samples (green). Additionally, concentrated grouping of QC runs in blue (Fig. 5), confirms method repeatability demonstrated in method validation.

The two clustered groups were analyzed using a supervised OPLS-DA, in order to find the discriminant features. The S-Plots obtained from OPLS-DA regression allowed to find potential discriminant features as shown in Table 2 and Additional file 1: Figures S5, S6, S7 and S8. Thus, in positive mode, one discriminant molecular feature  $(3.44_{-}612.1519n)$  was found in leaves samples, and another  $(3.47_{-}612.1522n)$  in roots samples. In negative mode, three discriminant molecular features  $(0.80_{-}131.0456 \text{ } m/z, 0.81_{-}114.0193 \text{ } m/z)$  and  $1.62_{-}191.0190 \text{ } m/z)$  were found in roots samples. All these features showed the highest variation. Thus, both in-house and online Kegg data-base were used to search and identify these features based on the exact mass, standard retention time (RT), and MS<sup>E</sup> spectra.

The two discriminant positive mode molecular features (3.44\_612.1519n and 3.47\_612.1522n) found in leaves and roots samples respectively, were identified as the sulfur metabolite glutathione oxidized (GSSG; see S-Plots in the Additional file 1: Figures S5 and S6). Identification was performed using exact mass, the RT of the standard reference and the MS/MS profile (level 1 of identification confidence [45]). Relative abundance obtained from Progenesis QI showed a low concentration of GSSG in the sulfur stressed group for both leaves and roots (Additional file 1: Figures S9 and S10). This metabolite was quantified using the calibration curve. Targeted quantification results represented in Fig. 6a revealed a coherence with relative quantification. GSSG concentration was notably lower in sulfur stressed plants, which was well explained by the sulfur deficiency in the literature [7, 8]. In fact, glutathione is a regulator of sulfur-uptake and assimilation. Hence, when the plant is sulfur-starved, the decrease of this compound increase transporter activity and maximize sulfate uptake [46].

In silico fragmentations from MS<sup>E</sup> acquisition of GSSG were also additional information for identification (Additional file 1: Figure S11).

Two metabolites were determined at level 1 in negative mode with OPLS-DA in roots, corresponding to the citric acid as  $1.62\_191.0190 \text{ m/z}$  and aspartic acid as  $0.81\_114.0193 \text{ m/z}$  (Additional file 1: Figure S7). Their relative abundance was correlated with the targeted quantification as shown in Fig. 6b and Additional file 1: Figures S12 and S13. As found in the quantification, the



on	
zati	
oni	
vei	
ati	
neg	
пр	
e a	
itiv	
bos	
2.	
ysis	
naly	
d ai	
ete	
arg	
unt	
Ţ	
s V	
ple	
am	
ts s	
õõ	
pu	
is al	
ave	
n le	
i p	
uno	
es fe	
tur	
fea	
lar.	
ecu	
lo	
nt n	
inal	
ini	
lisci	
/e d	
ativ	
put	
pu	
da	
ifie	
ent	
ldent	
ole 2 Ident	de

Predicted metabolite         m2         Formula         Mass error         Retention         Arrowa (p)         Pathway <i>Earlies</i> ESH         Metryl (15.57.35) -3 -hydroxy -2 -metryl/hexahydro-         217.154 $C_{H}$ / <sub>1</sub> /MO <sub>3</sub> -0.9         3.64         1.85E-12         Alkalodds <i>Herryl (15.57.35) -2 -hydroxy -2 -metryl/hexahydro-         217.154         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -1.9         3.64         1.85E-0.69         Valands           <i>Herryl (15.57.35) -2 -hydroxy -2 -metryl/hexahydro-         271.154         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -2.8         3.40         3.75E-06         Valands           <i>Herryl (15.57.35) -2 -hydroxy -2 -metryl/hexahydro-         273.134         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -2.1         3.40         3.75E-06         Valands           <i>Hydroxy -2 -metryl/hexahydro-         307.0332         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -0.2         3.44         3.55E-0.65         Sulfur metric           <i>(17.567.45) -3 (hydroxy-2 - metryl/hexahydro-         307.0332         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -0.2         3.41         3.55E-0.65         Sulfur metric           <i>(17.567.576) -3 (hydroxy-2 - metryl/hexahydro-         307.0332         <math>C_{H}</math>/<sub>1</sub>/MO<sub>3</sub>         -0.2         3.41         3.55E-0.65         Sulfur metric           <i>(17.567.576) -3 (hydrox-2 - metryl)         3.07.0332       </i></i></i></i></i></i></i>								
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	Predicted metabolite	z/m	Formula	Mass error (ppm)	Retention time (min)	Anova ( <i>p</i> )	Pathway	Level of identification confidence <sup>a</sup>
Methyl (15,25,735)-2 hydroxy-2-methylhrearhydro- 1H-pprofizine - lacitoxylare         217.1543 $C_{e}H_{1}NO_{7}$ -09         3.64         1355-12         Alkalodis d Alkalodis d - Svainsonine         1351 (17,25,735)-2 hydroxy-2-methylhrearhydro         1351 (21) $C_{e}H_{1}NO_{7}$ -09         3.44         1355-09         Valiance           (-)-Sonisonine         (2.5)-4 (beta-c-Glucopyranosyloxy)-2 (hydroxymethyl)-2         276.1073 $C_{H}H_{2}O_{1}S_{2}$ 0.2         3.44         1355-08         Valiance           (-)-Sonisonine         307.0332 $C_{a}H_{3}NO_{1}S_{2}$ 0.2         3.44         1555-08         Valiance           (1)-Soli-efficie         307.0334 $C_{a}H_{3}NO_{1}S_{2}$ 0.4         3.47         3.46         1555-08         Valiance           (1)-Soli-efficie         307.0334 $C_{a}H_{3}NO_{2}$ 0.2         3.44         1555-08         Valiance           (1)-Soli-efficie         307.0334 $C_{a}H_{3}NO_{2}$ 0.2         3.47         1355-03         Valiance           (1)-Soli-efficie         307.0334 $C_{a}H_{3}NO_{2}$ 0.2         3.47         3565-08         Valiance           (1)-Soli-efficie         307.0334 $C_{a}H_{3}NO_{2}$ 0.2         3.47         3565-0	Leaves—ESI+							
	Methyl (15,25,7aS)-2-hydroxy-2-methylhexahydro- 1H-pyrrolizine-1-carboxylate	217.1543	$C_{10}H_{17}NO_3$	- 0.9	3.64	1.85E—12	Alkaloids derived from ornithine	Level 4
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	(–)-Swainsonine	156.1021	C <sub>8</sub> H <sub>15</sub> NO <sub>3</sub>	1.1	3.91	6.98E-11	Alkaloids derived from ornithine	Level 4
Alphar-glucopyranosyl-         5431324 $C_{14}h_{3}N_{0}O_{12}^{2}$ 0.5         565E-08         Suffurmet           (1-86)-glucopyranosyl-         3070832 $C_{24}h_{3}N_{0}O_{12}^{2}$ -0.2         3.44         155E-05         Suffurmet           (1-86)-glucopyranose         3070834 $C_{24}h_{3}N_{0}O_{12}^{2}$ 0.4         3.47         736E-10         Suffurmet           Roust-field         3070834 $C_{34}h_{3}N_{0}O_{12}^{2}$ 0.4         3.47         736E-10         Suffurmet           Roust-field         1880710 $C_{14}h_{3}N_{0}O_{12}^{2}$ 0.4         3.47         736E-10         Suffurmet           Roust-field         1880710 $C_{14}h_{3}N_{0}O_{1}^{2}$ 4.7         3.55         Suffurmet           61         7 $C_{14}h_{3}N_{0}O_{1}^{2}$ 4.7         3.58         4.4E-08         Pineryptot           61         1880710 $C_{14}h_{3}N_{0}^{2}O_{1}^{2}$ 4.7         3.58         4.4E-08         Pineryptot           61         7 $C_{14}h_{3}O_{2}^{2}$ 4.7         3.85         1.44         1.60         2.41           7.5         6         7.6 $C_{14}h_{3}O_{2}^{2}$ 3.60         6         <	(2E)-4-(beta-D-Glucopyranosyloxy)-2-(hydroxymethyl)-2- butenenitrile	276.1073	$C_{11}H_{17}NO_7$	- 2.8	3.40	3.73E-09	Cyanogenic glucosides derived from valine or isoleucine	Level 4
Glutathione disulfide $3070832$ $C_{29}H_{32}N_{0}O_{15}S_{2}$ $-0.2$ $3.44$ $15F=05$ Sulfur methone disulfide <i>Roots-ESH</i> $80063-ESH$ $3070834$ $C_{29}H_{32}N_{0}O_{15}S_{2}$ $0.4$ $3.47$ $736F=10$ $8untanous disulfide           Roots-ESH         3070834 C_{39}H_{32}N_{0}O_{15}S_{2} 0.4 3.47 736F=10 8untanous disulfice           Roots-ESH         14704M_{3}S_{10} 14704M_{3}S_{10} 4.7 3.88 4.88E=00 Phenylorot Pc-Coumaric acid 14704M_{3}S_{10} 242.1748 C_{19}H_{30}O_{2} -0.9 3.88 4.48E=0.6 Phenylorot Acenosine S'-monphosphate 3165546 C_{11}H_{30}O_{2}S_{1} -3.4 3.10 2001502801 Farty acid           Acenosine S'-monphosphate         30.102545 C_{11}H_{30}O_{2}S_{2} 4.46+36 3.10 2001502801 Farty acid           Acenosine S'-monphosphate         30.102545 C_{11}H_{30}O_{2}S_{2} 3.4 4.16 8.666-0 9.486-06 9.486-06 9.486-06 9.486-06 $	Alpha-p-glucopyranosyl-(1->6)-alpha-p-glucopyranosyl- (1->6)-p-glucopyranose	543.1324	C <sub>18</sub> H <sub>32</sub> O <sub>16</sub>	0.5	1.23	5.65E-08		Level 4
Roots-ESH         Roots-ESH         307.0834 $C_{29}H_{3}N_{0}O_{1}S_{2}$ 0.4         3.47         7.36E-10         Suffurmet indication indication is indicating indicating indicating indicating is individual.           Indiotecrylic acid         147.0448 $C_{11}A_{3}N_{0}$ 1.0         3.62         8.865-10         Plenylproje indicating is indicating indicating indicating indicating indicating is individual.           A (5-Hydrowy-2-methyl-2-azabicyclo[2.2.2]oct-5-yl)         242.1748 $C_{11}A_{1}N_{0}O_{2}$ -0.9         3.81         0.001502861         Fatty acids individual individua individual individual individua individual individua indidua indidua individua individua indidua individua individua in	Glutathione disulfide	307.0832	C <sub>20</sub> H <sub>32</sub> N <sub>6</sub> O <sub>12</sub> S <sub>2</sub>	-0.2	3.44	1.55E-05	Sulfur metabolism	Level 1
Glurathione disulfide         307.0834 $C_{20}H_{3}N_{0}O_{12}S_{2}$ 0.4         3.47         7.36E-10         Sulfurmet:           indoleacrylic acid         18.0710 $C_{11}H_{3}N_{0}O_{2}$ 4.7         3.62         8.86E-10         Plant grow           P-Coumarie acid         147.0448 $C_{13}H_{23}N_{0}O_{2}$ 4.7         3.53         8.48E-10         Plant grow           4-6+Hydroxy-2-methyl-2-azabicydol2.2.2]oct-5-yl)-         242.1748 $C_{13}H_{23}N_{0}O_{2}$ -0.9         3.81         0.001502601         Fatry acids           3-methyllutanoic acid         147.0495         4.5         0.80         2.42E-14         Glucosinol           3-methyllutanoic acid         301.0545 $C_{11}H_{19}N_{0}O_{5}$ 4.5         0.80         2.42E-14         Glucosinol           3-5-2/Beta-p-glucopyranosyloxyl-3-methylbutanenitrile         301.1384 $C_{11}H_{19}N_{0}O_{6}$ -4.4         4.16         8.56E-09         Cyanogeni           35-2/Beta-p-glucopyranosyloxyl-3-methylbutanenitrile         301.138 $C_{11}H_{19}N_{0}O_{6}$ -4.4         4.16         8.56E-09         Cyanogeni           35-2/Beta-p-glucopyranosyloxyl-3-methylbutanenitrile         301.138 $C_{11}H_{19}N_{0}O_{6}$ -4.4         4.16         8.56E-09 </td <td>Roots—ESI+</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td>	Roots—ESI+							
Indoleacrylic acid         1880710 $C_{11}H_yNO_2$ 1.0         3.62         8.86E-10         Plant grow $P_{12}Mooxy-2$ -methyl-2-azabicyclo[2.22]oct-5-yl)-         147.0448 $C_{2}H_yO_2$ 4.7         3.58         4.48E-08         Phenylproj $-17.0448$ $C_{2}H_yO_2$ 3.10         2.001502861         Farty acids $-3-methyl-2-azabicyclo[2.22]oct-5-yl)-         2.42.1748         C_{12}H_{12}NO_2         3.81         0.001502861         Farty acids           -4merylputanoic acid         301.0545         C_{11}H_{20}O_5         3.3         0.001502861         Farty acids           2-Plydroxy-3-(6)-monphosphate         301.0545         C_{11}H_{20}O_5         3.3         0.011         3.4         3.10           2-Plydroxy-3-(6)-monphosphate         301.0545         C_{11}H_{10}O_5         3.9         0.90         5.43E-11         Flavonoid           5-Dliydroxy-4-ox-2-phenyl-7-chromanolate         301.0545         C_{11}H_{10}O_5         3.9         0.90         5.43E-11         Flavonoid           5-Dliydroxy-4-ogucopynanosyloxy)-3-methyl-3-butenen         301.0545         C_{11}H_{10}O_5         3.4         4.86         6.9         9.87E-08         Mmino aci           (25)-2-(8eta-0-glucopynanosyloxy)-3-methyl)-1-1(4+1$	Glutathione disulfide	307.0834	C <sub>20</sub> H <sub>32</sub> N <sub>6</sub> O <sub>12</sub> S <sub>2</sub>	0.4	3.47	7.36E-10	Sulfur metabolism	Level 1
p-Coumaric acid         147.0448 $C_9H_0O_3$ 4.7         3.58         44.8E-0.08         Phenylprogram $4.5$ -Hydroxy-2-methyl-2-azabicydo[2.2.2]oct-5-yl)- $3.21.1748$ $C_{13}H_{23}NO_3$ $-0.9$ $3.81$ 0.001502861         Faty acids $4.6$ -Hydroxy-2-methyl-2-azabicydo[2.2.2]oct-5-yl)- $242.1748$ $C_{13}H_{23}NO_3$ $-0.9$ $3.81$ $0.001502861$ Faty acids $4.6$ -methylsuffinyllberyllsuccinic acid $301.0545$ $C_{11}H_{13}O_5$ $4.5$ $0.80$ $2.422-14$ Glucosinol. $5-10hydroxy-4-oxo-2-phenyl-7-chromanolate         601.1384 C_{11}H_{13}NO_6 -4.4 3.57 1.80E-09 Cyanogeni licuic           5-25-2^{1}0erao-glucopyranosyloxy)-3-methylbutanentrile         304.1028 C_{11}H_{13}NO_6 -4.4 3.57 1.80E-09 Cyanogeni licuic           (25)-2^{-0}0erao-glucopyranosyloxy)-3-methylbutanentrile         306.11382 C_{13}H_{13}NO_6 -4.4 3.57 1.80E-09 Cyanogeni licuic           (25)-2^{-0}0erao-glucopyranosyloxy)-3-methylbutanentrile         306.1382 C_{11}H_{10}NO_6 -4.4 4.16 8.50E-09 Cyanogeni licuic      $	Indoleacrylic acid	188.0710	C <sub>11</sub> H <sub>9</sub> NO <sub>2</sub>	1.0	3.62	8.86E-10	Plant growth hormone	Level 4
4/5-Hydroxy-2-metryl-2-azbicyclo[2.2.2]oct-5-yl)-       24.1748 $C_{13}H_{13}NO_3$ -0.9       3.81       0.001502861       Faty acids         3-metryl/butanoic acid       3-metryl/butanoic acid       310.545 $C_{10}H_{13}N_0$ -0.9       3.81       0.001502861       Faty acids         2-metryl/butanoic acid       310.545 $C_{10}H_{13}N_0$ -3.4       3.10       2.01E-14       Zeatin bios         2-Hydroxy-3-fer(metrylsuffaryl/hexyl]succinic acid       301.0545 $C_{11}H_{13}NO_5$ 4.5       0.80       2.42E-14       Glucosinol         3.5-Dilydroxy-4-oxo-2-pheryl-7-chronanolate       601.1384 $C_{11}H_{13}NO_5$ -4.4       3.57       1.80E-09       Gyanogeni         5.5-Dilydroxy-4-oxo-2-pheryl-7-chronanolate       304.1028 $C_{11}H_{13}NO_5$ -4.4       3.57       1.80E-09       Gyanogeni         5.5-Dilydroxy-1-oxo-2-pheryl-7-chronanolate       304.1028 $C_{11}H_{13}NO_5$ -4.4       4.16       8.50E-09       Gyanogeni         5.5-J2-(Betra-0-glucopyranosyloxy)-3-methylbutanenitrile       306.1186 $C_{11}H_{10}NO_5$ -4.4       4.16       8.50E-09       Gyanogeni         1.5.7-2-(Betra-0-glucopyranosyloxy)-3-methylbutanenitrile       306.1186 $C_{11}H_{20}O_5$ 3.5       3.54E-06       Havonoid	p-Coumaric acid	147.0448	C <sub>9</sub> H <sub>8</sub> O <sub>3</sub>	4.7	3.58	4.48E—08	Phenylpropanoid	Level 4
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	4-(5-Hydroxy-2-methyl-2-azabicyclo[2.2.2]oct-5-yl)- 3-methylbutanoic acid	242.1748	C <sub>13</sub> H <sub>23</sub> NO <sub>3</sub>	0.0 —	3.81	0.001502861	Fatty acids	Level 4
Adenosine 5'-monophosphate         346.0546 $C_{10}H_{14}N_{0}O_{7}P$ $-3.4$ $3.10$ $2.01E-14$ Zeatin bios           2-Hydroxy-3-[6-(methylsulfanyl)hexyl]succinic acid         301.0545 $C_{11}H_{20}O_{5}S$ $4.5$ 0.80 $2.42E-14$ Glucosinol           3.5-Dihydroxy-4-xxo-2-phenyl-7-chromanolate $601.1384$ $C_{13}H_{11}N_{05}^{-}$ $3.9$ $0.90$ $5.43E-11$ Flavonoid $(25)-2.(Beta-p-glucopyranosyloxy)-3-methyl-3-butenen-         304.1028 C_{11}H_{13}N_{06}^{-} -4.4 3.57 180E-09 Cyanogeni leucine           (25)-2.(Beta-p-glucopyranosyloxy)-3-methylbutanentrile         306.1186 C_{11}H_{13}N_{06}^{-} -4.4 3.57 180E-09 Cyanogeni eucine           (25)-2.(Beta-p-glucopyranosyloxy)-3-methylbutanentrile         306.1186 C_{11}H_{13}N_{06}^{-} -4.4 4.16 8.56E-09 Cyanogeni eucine           (25)-2.(Beta-p-glucopyranosyloxy)-3-methylbutanentrile         306.1188 C_{11}H_{13}N_{06}^{-} -4.4 4.16 8.56E-09 Cyanogeni eucine           (25)-2.(Beta-p-glucopyranoside         601.1382 C_{13}H_{10}N_{0}^{-} -4.4 4.16 8.56E-09         <$	LeavesESI							
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	Adenosine 5'-monophosphate	346.0546	$C_{10}H_{14}N_5O_7P$	- 3.4	3.10	2.01E-14	Zeatin biosynthesis	Level 4
$\begin{array}{llllllllllllllllllllllllllllllllllll$	2-Hydroxy-3-[6-(methylsulfanyl)hexyl]succinic acid	301.0545	C <sub>11</sub> H <sub>20</sub> O <sub>5</sub> S	4.5	0.80	2.42E-14	Glucosinolates biosynthesis	Level 4
$ \begin{array}{rcl} (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl - 3 - butenen \\ itrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl - 3 - butenen \\ itrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl butanenitrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl butanenitrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl butanenitrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl butanenitrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 3 - methyl butanenitrile \\ (25)^2 - (Beta - D - glucopyranosyloxy) - 1 - (4 - hydroxyphenyl) - 1 - (4 - hydroxyphenyl) - 3 - (4 - hydroxyphenyl) - 1 - (4 - hydroxyphenyl) - 3 - (2 - H) = 0 \\ - 1 - propen - 1 - olate \\ (12)^3 - (2 - D - lithydroxyphenyl) - 1 - (4 - hydroxyphenyl) - 3 - (2 - H) \\ - 1 - propen - 1 - olate \\ (5 - H_1) - (1 - H_1) - (1 - H_1) - (4 - h_2) - (5 - H_1) - (5 - H_2) \\ - 1 - propen - 1 - olate \\ (5 - H_1) - (1 - H_1) - (1 - H_1) - (1 - H_2) - $	3,5-Dihydroxy-4-oxo-2-phenyl-7-chromanolate	601.1384	$C_{15}H_{11}O_5^{-}$	3.9	06.0	5.43E-11	Flavonoid biosynthesis	Level 4
$ (25)-2-(Beta-0-glucopyranosyloxy)-3-methylbutanenitrile 306.1186 C_{11}H_{19}NO_6 -4.4 4.16 8.50E-09 Cyanogeni leucine Hordatine B-like compounds 625.3092 C_{29}H_{40}N_6O_5 -4.19 6.19 9.87E-08 Amino acti (12)-3-(2,4-Dihydroxyphenyl)-1-(4-hydroxyphenyl)-3-oxo- 601.1382 C_{15}H_{10}O_5 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 653.1404 C_{25}H_{26}O_{12} 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 61.1382 C_{15}H_{10}O_5 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 601.1382 C_{15}H_{10}O_5 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 601.1382 C_{15}H_{10}O_5 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 601.1382 C_{15}H_{10}O_5 3.6 0.97 3.54E-06 Flavonoid Hordatine B-like compounds 601.1382 C_{5}H_{26}O_{12} -0.5 5.43 1.83E-04 Isoflavonoi 6-0-(carboxyacetyl)-beta-0-glucopyranoside Roots-ESI- 2-0, 0.079 1.06-16 Glucosinol 16-0, 0.087 C_{1}H_{20}O_5 3.4 0.79 1.10E-16 Glucosinol 128.0347 C_{5}H_{9}NO_4 -4.0 0.88 1.110E-16 Arginine ar Dehydroalanine 114.0193 C_{4}H_{7}NO_4 -2.4 0.81 1.11E-16 Cysteine ar Duchydrolanine B-tic acid B-ti$	(25)-2-(Beta-D-glucopyranosyloxy)-3-methyl-3-butenen- itrile	304.1028	C <sub>11</sub> H <sub>17</sub> NO <sub>6</sub>	- 4,4	3.57	1.80E-09	Cyanogenic glucosides derived from leucine	Level 4
Hordatine B-like compounds $625.3092$ $C_{29}H_{40}N_6O_5$ $-4.9$ $6.19$ $987E-08$ Amino activation activ	(25)-2-(Beta-D-glucopyranosyloxy)-3-methylbutanenitrile	306.1186	C <sub>11</sub> H <sub>19</sub> NO <sub>6</sub>	-4.4	4.16	8.50E-09	Cyanogenic glucosides derived from leucine	Level 4
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	Hordatine B-like compounds	625.3092	C <sub>29</sub> H <sub>40</sub> N <sub>8</sub> O <sub>5</sub>	- 4.9	6.19	9.87E-08	Amino acid related compounds	Level 4
(6aR,11aP)-9-Methoxy-6a,11a-dihydro-       563.1404 $C_{25}H_{26}O_{12}$ -0.5       5.43       183E-04       Isoflavonoi         6H-[1]benzofund]3.2-clchromen-3-yl       60.2       5.41       1.83E-04       Isoflavonoi         6H-[1]benzofund]3.2-clchromen-3-yl       503.1404 $C_{25}H_{26}O_{12}$ -0.5       5.43       1.83E-04       Isoflavonoi         6-0-(carboxyacety))-beta-p-glucopyranoside       301.0547 $C_{11}H_{20}O_{5}S$ 3.4       0.79       1.10E-16       Glucosinol <i>Roots-ESI-</i> 2-Hydroxy-2-[6-(methylsulfanyl)hexyl]succinic acid       301.0547 $C_{11}H_{20}O_{5}S$ 3.4       0.79       1.10E-16       Arginine ar         Roots-ESI-       2-Hydroxy-2-[6-(methylsulfanyl)hexyl]succinic acid       128.0347 $C_{3}H_{9}NO_{4}$ -4.0       1.58       1.10E-16       Arginine ar         Dehydroalanine       260.0878 $C_{3}H_{5}NO_{2}$ -4.0       0.88       1.11E-16       Cysteine ar         Du-Aspartic acid       114.0193 $C_{4}H_{7}NO_{4}$ -2.4       0.81       1.11E-16       Amino acid	(1Z)-3-(2,4-Dihydroxyphenyl)-1-(4-hydroxyphenyl)-3-oxo- 1-propen-1-olate	601.1382	C <sub>15</sub> H <sub>11</sub> O <sub>5</sub> <sup>-</sup>	3.6	0.97	3.54E-06	Flavonoid biosynthesis	Level 4
2-Hydroxy-2-[6-(methylsulfanyl)hexyl]succinic acid 301.0547 $C_{11}H_{20}O_5S$ 3.4 0.79 1.10E-16 Glucosinol. (4R)-5-Amino-4-hydroxy-2-oxopentanoic acid 128.0347 $C_5H_9NO_4$ -4.0 1.58 1.10E-16 Arginine ar Dehydroalanine 1.11E-16 $C_5H_9NO_2$ -4.0 0.88 1.11E-16 Cysteine ar DL-Aspartic acid 114.0193 $C_4H_7NO_4$ -2.4 0.81 1.11E-16 Amino acid DL-Aspartic acid 1.11E-16 $C_5H_9NO_4$ -2.4 0.81 0.81 0.81 0.81 0.81 0.81 0.81 0.81	(6aR, 11 aR)-9-Methoxy-6a,11a-dihydro- 6H-[1]benzofuro[3,2-c]chromen-3-yl 6-O-(carboxyacetyl)-beta-p-glucopyranoside <i>Roots—ESI</i> -	563.1404	C <sub>25</sub> H <sub>26</sub> O <sub>12</sub>	- 0.5	5.43	1.83E04	Isoflavonoid biosynthesis	Level 4
$ \begin{array}{llllllllllllllllllllllllllllllllllll$	2-Hvdroxv-2-[6-(methylsulfany]]hexyl]surcrinic acid	301 0547	S-O-H-C	78	0 79	1 10E—16	Glucosinalates biosvathesis	ן פועפן ב
Dehydroalanine 260.0878 $C_3H_5NO_2$ -4.0 0.88 1.11E-16 Cysteine ar DL-Aspartic acid 114.0193 $C_4H_5NO_4$ -2.4 0.81 1.11E-16 Amino acid	(4R)-5-Amino-4-hvdroxy-2-oxopentanoic acid	128.0347	C <sub>E</sub> H <sub>6</sub> NO 4	- 4.0	1.58	1.10E-16	Arainine and proline metabolism	Level 4
DL-Aspartic acid 114.0193 C <sub>4</sub> H <sub>7</sub> NO <sub>4</sub> – 2.4 0.81 1.11E–16 Amino acid	Dehydroalanine	260.0878	C <sub>3</sub> H <sub>5</sub> NO <sub>2</sub>	- 4.0	0.88	1.11E-16	Cysteine and methionine metabolism	Level 4
	DL-Aspartic acid	114.0193	$C_4H_7NO_4$	- 2.4	0.81	1.11E-16	Amino acids	Level 1
L-(+)-Asparagine	L-(+)-Asparagine	131.0456	$C_4H_{10}N_2O_4$	- 4.1	0.80	4.22E-15	Amino acids	Level 1

Predicted metabolite	m/z	Formula	Mass error (ppm)	Retention time (min)	Anova ( <i>p</i> )	Pathway	Level of identification confidence <sup>a</sup>
5-Hydroxy-2-(3-hydroxy-4-methoxyphenyl)-4-oxo-7- chromanolate	601.1384	C <sub>16</sub> H <sub>13</sub> O <sub>6</sub> <sup>-</sup>	3.6	06.0	5.22E-15	Flavonones biosynthesis	Level 4
Benzoyl-beta-p-glucoside	329.0866	$C_{13}H_{16}O_7$	-4.4	4.66	2.44E—14	Glycosyl ester	Level 4
Citric acid	191.0190	C <sub>6</sub> H <sub>10</sub> O <sub>8</sub>	- 3.3	1.62	5.17E-13	TCA cycle	Level 1
L-Glutamine	145.0612	$C_{5}H_{10}N_{2}O_{3}$	-4.2	0.83	5.78E-12	Biosynthesis of amino acids	Level 1
Sinapic acid	245.0423	C <sub>11</sub> H <sub>12</sub> O <sub>5</sub>	- 3.7	0.98	6.99E-12	Phenylpropanoids	Level 4
Oxidized glutathione	611.1441	C <sub>20</sub> H <sub>32</sub> N <sub>6</sub> O <sub>12</sub> S <sub>2</sub>	- 1.1	2.05	8.69E—09	Sulfur metabolism	Level 1
These molecular features were considered as discriminant usi	ng OPLS-DA. Glu	tathione disulfide (G	SSG) was identil	îed in positive i	onization with [N	1+2HJ <sup>2+</sup> ion	

Table 2 (continued)

<sup>a</sup> According to Schymanski et al. [45]





2–3 fold of aspartic acid increasing was mentioned by Zhao et al. [40] as one of sulfur deficiency signs.

On the other hand, asparagine was detected and identified at level 1 as  $0,80\_131,0456 \text{ m/z}$  in negative mode in roots (Additional file 1: Figure S7) showing a high concentration in sulfur depleted plants while it showed a very low concentration in sulfur sufficient plants (Additional file 1: Figure S14). In fact, asparagine and arginine act as primary and secondary storage of nitrogen respectively in sulfur-depleted plants, as demonstrated by Mertz et al. [47]. Thus, targeted quantification of asparagine and arginine in ESI+ is shown in Fig. 6c, demonstrating a clear coherence with untargeted analysis of asparagine and biological explanation.

Hence, according to Schymanski et al. [45], several discriminant molecular features were identified with the level 1 of identification confidence (Table 2). This is by confirming the structure using comparisons with the RT and the MS and MS/MS spectra of reference standards. Other discriminant molecular features represented in Table 2 were identified with the level 4 of identification confidence, due to lack of standard references. The level 4 was reached by elemental compositions identification using exact mass, isotopic patterns, adducts and *in silico* fragmentations. This identification is provided

by the software algorithm (Progenesis QI). On the other hand, 1844 molecular features were found in roots (all in both positive and negative ionization modes) and 1573 molecular features were found in leaves (all in both positive and negative ionization modes) after application of the 0.05 p value and the  $\geq 2$  max fold change filters. 272 molecular features could be identified in roots and 342 molecular features in leaves using a barley-specified inhouse database. These metabolites were also identified with the level 4 of identification confidence. Otherwise, targeted metabolites quantification in roots is represented in Fig. 7.

## Validation in barley samples

To assess the targeted method of polar metabolite analysis, barley root samples (control batch) were analyzed with both methods. Thus, this method showed a good precision in retention time (RSD below 2%) for ten repeated injections for all detected amino acids and sulfur metabolites (Table 3). All 17 metabolites could be quantified with an intra-day RSD below 8% in comparison between 4 biological replicates (n=4) and an interday RSD (n=4) below 9% within 6 months (Table 3).

A comparable precision was obtained in negative mode. Retention time (RSD below 2.5%) showed a good

Table 3 Intra-sample, intra-day and inter-day validation

Compound	Intra-sample RSD (%) of RT	Intra-day RSD (%) of PA (n=4)	Inter-day RSD (%) of PA (n=4)
ESI—			
Fumaric acid	0.45	4.70	7.59
Succinic acid	0.34	5.61	8.77
Aspartic acid	0.00	2.42	5.46
Malic acid	0.45	6.97	9.83
Glutamic acid	0.00	5.41	7.07
Phospho(enol) pyruvic acid	2.48	4.88	9.97
Gallic acid	0.27	4.53	N.Q.
Cis-Aconitic acid	0.25	3.94	N.Q.
Shikimic acid	0.62	1.63	N.Q.
Azelaic acid	0.14	5.43	N.Q.
Citric acid	0.28	2.20	3.79
Isocitric acid	0.00	1.79	5.11
Gluconic acid	0.00	4.05	3.51
Kaempferol	0.07	4.70	N.Q.
Chlorogenic acid	0.17	4.27	N.Q.
Trehalose 6-phos- phate	0.00	6.37	5.34
ESI+			
Proline	0.42	4.02	1.91
Isoleucine	0.46	0.98	3.82
Leucine	0.20	2.32	7.24
Asparagine	0.00	7.024	7.91
Glutamine	0.00	2.32	4.30
Lysine	0.59	4.82	7.28
O-Acetyl-serine	0.96	2.74	4.90
Methionine	0.35	2.95	8.08
Histidine	0.00	3.32	2.85
Phenylalanine	0.13	2.02	3.81
Arginine	0.00	3.89	3.76
Tyrosine	0.35	1.26	7.14
Tryptophan	0.00	4.85	4.60
Thiamine	0.77	5.82	4.11
Glutathione reduced	0.52	6.09	8.94
S-adenosyl-methio- nine	1.13	5.95	5.78
Glutathione oxi- dized	0.21	2.48	8.72

Sample RSD (%) of RT was calculated from ten repeated injections of the same extract (replicate 1). Intra-day and inter-day RSD (%) were calculated from four biological replicates. Inter-day quantification was realized within 6 months

RT Retention time, PA peak area, N.Q. not quantified

precision for ten repeated injections for all detected organic acids, amino acids, phosphorylated sugars and secondary metabolites (Table 3). Intra-day quantification RSD (n = 4) was below 7%, and inter-day quantification

RSD (n = 4) was below 10%. However, gallic acid, cisaconitic acid, shikimic acid, kaempferol and chlorogenic acid were quantified near the LOQ, and were not quantified after 6 months due to a potential degradation in plant samples (Table 3).

Finally, the simultaneous untargeted/targeted method was successfully applied to real samples, demonstrating high reproducibility with a RSD values below 10%.

It is worth to mention that only 20 mg of fresh material were used to detect 33 underivatized primary metabolites with a high sensitivity and fast analysis at high resolution mass detection. Perchloric acid was added to the solvent in order to reduce the risk of degradation of sulfur containing metabolites. Moreover, untargeted analysis allowed to discriminate between sulfur depleted and controlled barley enabling the identification of several discriminant features related to the primary metabolism under stress conditions.

## Conclusions

A simultaneous untargeted/targeted UPLC-HRMS based method has been developed, providing complementary and reliable information within 7-10 min for a single run, allowing high-throughput analysis. Both UPLC HSS T3 and Luna® Omega PS C18 columns improved considerably retention and chromatographic resolution of polar compounds. The optimized chromatographic conditions allowed to separate 33 primary metabolites including isomers (isoleucine and leucine, isocitrate and citrate) without any derivatization or additional complex sampling step, allowing simple, rapid, and reproducible analysis of these metabolites, but also allowing untargeted metabolic profiling. On the other hand, high resolution mass spectrometry provided high selectivity for untargeted analysis. It also provides reliable and sensitive compound detection and quantification with accurate mass measurement in complex samples, which allowed to discriminate between compounds with the same nominal mass, potential co-eluted interferences, and isotopes contributions. The MS<sup>E</sup> data acquisition supplied a structural information that can be used for compound identification. The method has succeeded to discriminate between different plant batches under sulfur controlled/deficient, and allowed to identify several biomarkers confirmed by the untargeted/targeted profiling analysis. This work opens interesting perspectives in both fundamental and applied research. Indeed, biomarkers give precious indication on the mechanisms that govern the plant nutrition, especially during a nutritional deficiency. The development of decision support tools based on a direct or indirect measurement of these metabolites would be promising for the

plant nutritional status, thus allowing a real time fertilization management and encounter the challenges of sustainable agriculture.

## **Additional file**

Additional file 1. Table S1. RSD – ESI+ – Lowest concentration. Table S2. RSD – ESI+ – Intermediate concentration. Table S3. RSD – ESI+ – Highest concentration. Table S4. RSD – ESI- – Lowest concentration. Table S5. RSD – ESI- – Intermediate concentration. Table S6. RSD – ESI- – Highest concentration. Table S7. Adducts list. Table S8. Time limits. Figure S1. EIC m/z 173.0092. Figure S2. Isocitrate source fragmentation – MS spectrum. Figure S3. cis-Aconitic acid –  $C_6H_6O_6$ . Figure S4. Isocitric acid –  $C_6H_8O_7$ . Figure S5. S-Plot - Roots - ESI+. Figure S6. S-Plot - Leaves - ESI+. Figure S7. S-Plot - Roots - ESI+. Figure S10. GSSG relative abundance – Leaves. Figure S11. In silico fragmentations from MS<sup>E</sup> acquisition of GSSG ure S13. Aspartate relative abundance – Roots. Figure S14. Asparagine relative abundance – Roots. Figure S15. Extracted ion chromatograms.

#### Abbreviations

GC-MS: gas chromatography-mass spectrometry; LC-MS: liquid chromatography-mass spectrometry; HILIC: hydrophilic interaction liquid chromatography; HPLC-MS: high performance liquid chromatography-mass spectrometry; EI: electronic impact; UPLC-HRMS: ultra-high performance liquid chromatography-high resolution mass spectrometry; LC-HRMS: liquid chromatographyhigh resolution mass spectrometry; QToF MS: quadrupole/time-of-flight mass spectrometer; ESI: electrospray ionization; FWHM: full width half maximum; LOD: limit of detection; LOQ: limit of quantitation; RPM: rotation per minute; QC: quality control; PCA: principal components analysis; OPLS-DA: orthogonal projections to latent structures discriminant analysis; ANOVA: analysis of variance; S-Plots: score plots; RT: retention time; GSH: glutathione reduced; GSSG: glutathione oxidized; SAM: S-adenosyl-methionine.

#### Authors' contributions

AS, FJ and JCY designed the study. AS and HG developed and performed the UPLC–HRMS method, and drafted the manuscript. HG carried out the experimental work. All authors read and approved the final manuscript.

#### Author details

<sup>1</sup> Centre Mondial de l'Innovation Roullier (CMI), 18 Avenue Franklin Roosevelt, 35400 Saint-Malo, France. <sup>2</sup> UFR Sciences et Propriétés de la Matière, Université de Rennes 1, 2 rue du Thabor, CS 46510, 35065 Rennes Cedex, France.

#### Acknowledgements

The authors thank the «Fonds unique interministériel—FUI» which supported this work conducted through the SERAPIS project. Acknowledgments to Dr. Areej Merhi for providing English language editing, Dr. Mustapha Arkoun, Dr. Eric Nguema-Ona, and Dr. Nicolas Le Yondre for their scientific advice, Dr. Seyed Abdollah Hosseini and Victor Maignan for providing barley samples.

#### **Competing interests**

The authors declare that they have no competing interests.

#### Availability of data and materials

The datasets used and/or analyzed during the current study are available from the corresponding author on reasonable request.

#### **Consent for publication**

Not applicable.

#### Ethics approval and consent to participate

Not applicable.

#### Funding

This work was supported by the "Fonds unique interministériel – FUI" which supported this work conducted through the SERAPIS.

## Publisher's Note

Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Received: 10 April 2018 Accepted: 11 July 2018 Published online: 24 July 2018

#### References

- 1. Lei Z, Huhman DV, Sumner LW. Mass spectrometry strategies in metabolomics. J Biol Chem. 2011;286:25435–42.
- Jorge TF, Rodrigues JA, Caldana C, Schmidt R, van Dongen JT, Thomas-Oates J, et al. Mass spectrometry-based plant metabolomics: metabolite responses to abiotic stress. Mass Spectrom Rev. 2016;35:620–49. https:// doi.org/10.1002/mas.21449.
- Jorge TF, Duro N, da Costa M, Florian A, Ramalho JC, Ribeiro-Barros AI, et al. GC-TOF-MS analysis reveals salt stress-responsive primary metabolites in *Casuarina glauca* tissues. Metabolomics. 2017;13:1–13.
- Schäfer M, Brütting C, Baldwin IT, Kallenbach M. High-throughput quantification of more than 100 primary- and secondary-metabolites, and phytohormones by a single solid-phase extraction based sample preparation with analysis by UHPLC-HESI-MS/MS. Plant Methods. 2016;12:30. https://doi.org/10.1186/s13007-016-0130-x.
- Chia DW, Yoder TJ, Reiter WD, Gibson SI. Fumaric acid: an overlooked form of fixed carbon in Arabidopsis and other plant species. Planta. 2000;211:743–51.
- Sanchez DH, Siahpoosh MR, Roessner U, Udvardi M, Kopka J. Plant metabolomics reveals conserved and divergent metabolic responses to salinity. Physiol Plant. 2008;132:209–19.
- Carfagna S, Vona V, Di Martino V, Esposito S, Rigano C. Nitrogen assimilation and cysteine biosynthesis in barley: evidence for root sulphur assimilation upon recovery from N deprivation. Environ Exp Bot. 2011;71:18–24. https://doi.org/10.1016/j.envexpbot.2010.10.008.
- Zuber H, Poignavent G, Le Signor C, Aimé D, Vieren E, Tadla C, et al. Legume adaptation to sulfur deficiency revealed by comparing nutrient allocation and seed traits in *Medicago truncatula*. Plant J. 2013;76:982–96.
- Arrivault S, Guenther M, Ivakov A, Feil R, Vosloh D, Van Dongen JT, et al. Use of reverse-phase liquid chromatography, linked to tandem mass spectrometry, to profile the Calvin cycle and other metabolic intermediates in Arabidopsis rosettes at different carbon dioxide concentrations. Plant J. 2009;59:824–39.
- Erro J, Zamarreno AM, Yvin JC, Garcia-Mina JM. Determination of organic acids in tissues and exudates of maize, lupin, and chickpea by high-performance liquid chromatography-tandem mass spectrometry. J Agric Food Chem. 2009;57:4004–10.
- Rellán-Álvarez R, Hernández LE, Abadía J, Álvarez-Fernández A. Direct and simultaneous determination of reduced and oxidized glutathione and homoglutathione by liquid chromatography-electrospray/mass spectrometry in plant tissue extracts. Anal Biochem. 2006;356:254–64.
- Tolstikov VV, Fiehn O. Analysis of highly polar compounds of plant origin: combination of hydrophilic interaction chromatography and electrospray ion trap mass spectrometry. Anal Biochem. 2002;301:298–307.
- 13. Liu Z, Rochfort S. A fast liquid chromatography–mass spectrometry (LC–MS) method for quantification of major polar metabolites in plants. J Chromatogr B Anal Technol Biomed Life Sci. 2013;912:8–15. https://doi.org/10.1016/j.jchromb.2012.10.040.
- Chang YL, Hsieh CL, Huang YM, Chiou WL, Kuo YH, Tseng MH. Modified method for determination of sulfur metabolites in plant tissues by stable isotope dilution-based liquid chromatography–electrospray ionization–tandem mass spectrometry. Anal Biochem. 2013;442:24–33. https://doi.org/10.1016/j.ab.2013.07.026.
- 15. Guan X, Hoffman B, Dwivedi C, Matthees DP. A simultaneous liquid chromatography/mass spectrometric assay of glutathione, cysteine,

homocysteine and their disulfides in biological samples. J Pharm Biomed Anal. 2003;31:251–61.

- Le A, Ng A, Kwan T, Cusmano-Ozog K, Cowan TM. A rapid, sensitive method for quantitative analysis of underivatized amino acids by liquid chromatography-tandem mass spectrometry (LC–MS/MS). J Chromatogr B Anal Technol Biomed Life Sci. 2014;944:166–74. https:// doi.org/10.1016/j.jchromb.2013.11.017.
- Horňák K, Schmidheiny H, Pernthaler J. High-throughput determination of dissolved free amino acids in unconcentrated freshwater by ion-pairing liquid chromatography and mass spectrometry. J Chromatogr A. 2016;1440:85–93.
- Hammermeister DE, Serrano J, Schmieder P, Kuehl DW. Characterization of dansylated glutathione, glutathione disulfide, cysteine and cystine by narrow bore liquid chromatography/electrospray ionization mass spectrometry. Rapid Commun Mass Spectrom. 2000;14:503–8.
- Dziagwa-Becker MM, Ramos JMM, Topolskia JK, Oleszek WA. Determination of free amino acids in plants by liquid chromatography coupled to tandem mass spectrometry (LC–MS/MS). Anal Methods. 2015;7:7574–81. https://doi.org/10.1016/j.jchromb.2015.10.040.
- Camera E, Rinaldi M, Briganti S, Picardo M, Fanali S. Simultaneous determination of reduced and oxidized glutathione in peripheral blood mononuclear cells by liquid chromatography–electrospray mass spectrometry. J Chromatogr B Biomed Sci Appl. 2001;757:69–78.
- Dell'mour M, Jaitz L, Oburger E, Puschenreiter M, Koellensperger G, Hann S. Hydrophilic interaction LC combined with electrospray MS for highly sensitive analysis of underivatized amino acids in rhizosphere research. J Sep Sci. 2010;33:911–22.
- Mathon C, Barding GA, Larive CK. Separation of ten phosphorylated mono-and disaccharides using HILIC and ion-pairing interactions. Anal Chim Acta. 2017;972:102–10. https://doi.org/10.1016/j.aca.2017.03.029.
- t'Kindt R, Morreel K, Deforce D, Boerjan W, Van Bocxlaer J. Joint GC–MS and LC–MS platforms for comprehensive plant metabolomics: repeatability and sample pre-treatment. J Chromatogr B Anal Technol Biomed Life Sci. 2009;877:3572–80.
- Lisec J, Schauer N, Kopka J, Willmitzer L, Fernie AR. Gas chromatography mass spectrometry-based metabolite profiling in plants. Nat Protoc. 2006;1:387–96.
- Zhen S, Zhou J, Deng X, Zhu G, Cao H, Wang Z, et al. Metabolite profiling of the response to high-nitrogen fertilizer during grain development of bread wheat (*Triticum aestivum* L.). J Cereal Sci. 2016;69:85–94. https://doi. org/10.1016/j.jcs.2016.02.014.
- Richter JA, Erban A, Kopka J, Zörb C. Metabolic contribution to salt stress in two maize hybrids with contrasting resistance. Plant Sci. 2015;233:107– 15. https://doi.org/10.1016/j.plantsci.2015.01.006.
- 27. Obata T, Fernie AR. The use of metabolomics to dissect plant responses to abiotic stresses. Cell Mol Life Sci. 2012;69:3225–43.
- Bedair M, Sumner LW. Current and emerging mass-spectrometry technologies for metabolomics. TrAC Trends Anal Chem. 2008;27:238–50. https://doi.org/10.1016/J.TRAC.2008.01.006.
- Drapal M, Farfan-Vignolo ER, Gutierrez OR, Bonierbale M, Mihovilovich E, Fraser PD. Identification of metabolites associated with water stress responses in *Solanum tuberosum* L. clones. Phytochemistry. 2017;135:24– 33. https://doi.org/10.1016/j.phytochem.2016.12.003.
- Baniasadi H, Vlahakis C, Hazebroek J, Zhong C, Asiago V. Effect of environment and genotype on commercial maize hybrids using LC/MS-based metabolomics. J Agric Food Chem. 2014;62:1412–22.
- Chamam A, Sanguin H, Bellvert F, Meiffren G, Comte G, Wisniewski-Dyé F, et al. Plant secondary metabolite profiling evidences strain-dependent effect in the *Azospirillum–Oryza sativa* association. Phytochemistry. 2013;87:65–77. https://doi.org/10.1016/j.phytochem.2012.11.009.
- 32. t'Kindt R, De Veylder L, Storme M, Deforce D, Van Bocxlaer J. LC–MS metabolic profiling of *Arabidopsis thaliana* plant leaves and cell cultures: Optimization of pre-LC–MS procedure parameters. J Chromatogr B Anal Technol Biomed Life Sci. 2008;871:37–43.
- 33. Grata E, Guillarme D, Glauser G, Boccard J, Carrupt PA, Veuthey JL, et al. Metabolite profiling of plant extracts by ultra-high-pressure liquid

chromatography at elevated temperature coupled to time-of-flight mass spectrometry. J Chromatogr A. 2009;1216:5660–8.

- Liu H, Lai H, Jia X, Liu J, Zhang Z, Qi Y, et al. Comprehensive chemical analysis of *Schisandra chinensis* by HPLC–DAD–MS combined with chemometrics. Phytomedicine. 2013;20:1135–43. https://doi.org/10.1016/j. phymed.2013.05.001.
- 35. Zhou Y, Huang S-X, Pu J-X, Li J-R, Ding L-S, Chen D-F, et al. Ultra performance liquid chromatography coupled with quadrupole time-of-flight mass spectrometric procedure for qualitative and quantitative analyses of nortriterpenoids and lignans in the genus *Schisandra*. J Pharm Biomed Anal. 2011;56:916–27. https://doi.org/10.1016/j.jpba.2011.07.045.
- García-Villalba R, Tomás-Barberán F, Fança-Berthon P, Roller M, Zafrilla P, Issaly N, García-Conesa M-T. Targeted and untargeted metabolomics to explore the bioavailability of the secoiridoids from a seed/fruit extract (*Fraxinus angustifolia* Vahl) in human healthy volunteers: a preliminary study. Molecules. 2015;20:22202–19. https://doi.org/10.3390/molecules2 01219845.
- Mechri B, Tekaya M, Cheheb H, Attia F, Hammami M. Accumulation of flavonoids and phenolic compounds in olive tree roots in response to mycorrhizal colonization: a possible mechanism for regulation of defense molecules. J Plant Physiol. 2015;185:40–3. https://doi.org/10.1016/j.jplph .2015.06.015.
- Treutter D. Significance of flavonoids in plant resistance and enhancement of their biosynthesis. Plant Biol. 2005;7:581–91.
- Zhao FJ, Hawkesford MJ, Warrilow AGS, McGrath SP, Clarkson DT. Responses of two wheat varieties to sulphur addition and diagnosis of sulphur deficiency. Plant Soil. 1996;181:317–27. https://doi.org/10.1007/ BF00012066.
- Zhao FJ, Hawkesford MJ, Mcgrath SP. Sulphur assimilation and effects on yield and quality of wheat. J Cereal Sci. 1999;30:1–17. https://doi. org/10.1006/jcrs.1998.0241.
- Kruger NJ, Troncoso-Ponce MA, Ratcliffe RG. 1H NMR metabolite fingerprinting and metabolomic analysis of perchloric acid extracts from plant tissues. Nat Protoc. 2008;3:1001–12. https://doi.org/10.1038/nprot .2008.64.
- Lafaye A, Labarre J, Tabet J, Ezan E, Junot C, De Pharmacologie S, et al. Liquid chromatography–mass spectrometry and N metabolic labeling for quantitative metabolic profiling. Anal Chem. 2005;77:2026–33.
- Hsieh CL, Yeh KW, De Kok LJ, Pan RN, Kuo YH, Tseng MH. Simultaneous determination of sulphur metabolites in *Arabidopsis thaliana* via LC-ESI-MS/MS and 34S-metabolic labelling. Phytochem Anal. 2012;23:324–31.
- Kellmann M, Muenster H, Zomer P, Mol H. Full scan MS in comprehensive qualitative and quantitative residue analysis in food and feed matrices: how much resolving power is required? J Am Soc Mass Spectrom. 2009;20:1464–76. https://doi.org/10.1016/J.JASMS.2009.05.010.
- Schymanski EL, Jeon J, Gulde R, Fenner K, Ruff M, Singer HP, et al. Identifying small molecules via high resolution mass spectrometry: communicating confidence. Environ Sci Technol. 2014;48:2097–8.
- Hoefgen R. Molecular analysis and control of cysteine biosynthesis: integration of nitrogen and sulphur metabolism. J Exp Bot. 2004;55:1283–92. https://doi.org/10.1093/jxb/erh136.
- Mertz ET, Singleton VL, Garey CL. The effect of sulfur deficiency on the amino acids of alfalfa. Arch Biochem Biophys. 1952;38:139–45. https://doi. org/10.1016/0003-9861(52)90017-9.

#### Ready to submit your research? Choose BMC and benefit from:

- fast, convenient online submission
- thorough peer review by experienced researchers in your field
- rapid publication on acceptance
- support for research data, including large and complex data types
- gold Open Access which fosters wider collaboration and increased citations
- maximum visibility for your research: over 100M website views per year

#### At BMC, research is always in progress.

Learn more biomedcentral.com/submissions

